doi:10.2533/chimia.2014.239

Synthesis and Characterization of Photoaffinity Probes that Target the 5-HT₃ Receptor

Thomas Jack[§], Marc-David Ruepp, Andrew J. Thompson, Oliver Mühlemann, and Martin Lochner*

§SCS-Metrohm Foundation Award for best oral presentation

Abstract: The 5-HT₃ receptor is one of several ion channels responsible for the transmission of nerve impulses in the peripheral and central nervous systems. Until now, it has been difficult to characterize transmembrane receptors with classical structural biology approaches like X-ray crystallography. The use of photoaffinity probes is an alternative approach to identify regions in the protein where small molecules bind. To this end, we present two photoaffinity probes based on granisetron, a well known antagonist of the 5-HT₃ receptor. These new probes show nanomolar binding affinity for the orthosteric binding site. In addition, we investigated their reactivity using irradiation experiments.

Keywords: Granisetron \cdot 5-HT₃ receptor \cdot Photoaffinity probes \cdot Photo-labeling \cdot Serotonin receptor

Introduction

In popular science serotonin is referred as a 'happiness hormone'. Also known as 5-hydroxytryptamine (5-HT), it is an important neurotransmitter that is involved in many processes in the human body such as the regulation of appetite, mood, sleep, vasoconstriction, body temperature and gastrointestinal motility. It may also be associated with disorders such as aggression, anxiety, addiction, emesis and nausea.^[1]

The family of human serotonin receptors contains seven different subtypes (5- HT_1 to 5- HT_7) that are widely distributed throughout the body.^[2] Most serotonin receptors are G-protein coupled receptors that influence the intracellular concentrations of cAMP and IP3.^[3] The only exception is the 5- HT_3 receptor, a ligand-gated ion channel selective for sodium and potassium ions. They are transmembrane proteins, composed of five subunits that

surround an ion-conducting pore and their activation triggers a neuronal excitation by depolarization of the cell plasma membrane (Fig. 1).^[4]

There are several approved drugs that selectively antagonize 5-HT₃ receptors and these are primarily used to treat chemo-therapy-, radiotherapy- and post operative-induced nausea and vomiting.

Binding of serotonin triggers the opening of an ion channel, but competitive antagonists like granisetron are able to block the orthosteric binding site and thereby inhibit pore opening. Until now, a high-resolution crystal structure of the 5-HT₃ receptor has not been obtained and our knowledge of the binding cavity has largely been based on mutagenesis and homology with related proteins.^[5]

The aim of our project is the development of photoaffinity probes that can be used to covalently modify the 5-HT₃ receptor, and after subsequent enzymatic digestion, identify the orientation of the ligand in the binding pocket by mass spectrometry (Scheme 1).

We base our probe design on the known high-affinity ligand granisetron.^[6] Previous studies in our group have identified three positions where bulky groups can be introduced to the molecule without losing the binding affinity of the parent compound (Fig. 2).^[7]

Synthesis of Photolabile Antagonists

Based on the structure–activity relationship studies of granisetron we introduced a photolabile diazirine at the indazole position N(1) and C(7) to obtain two final compounds **1** and **2**, which were synthesized according to the procedure shown in Schemes 2–5.

The compounds 1 and 2 were assem-

*Correspondence: Prof. Dr. M. Lochner Universität Bern Departement für Chemie und Biochemie Freiestrasse 3 CH-3012 Bern Tel.: +41 31 631 3311 E-mail: martin.lochner@dcb.unibe.ch

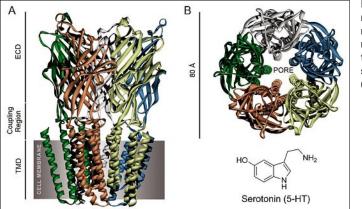
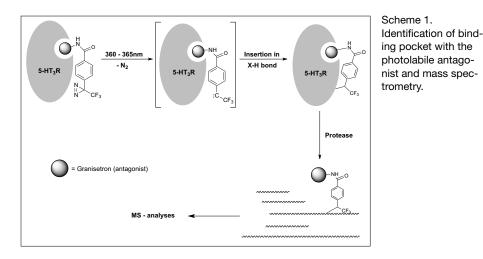


Fig. 1. Homology model of the 5-HT $_3$ receptor based on a related GLIC structure (PDB ID: 3EAM) showing the side view (A) and top view (B).



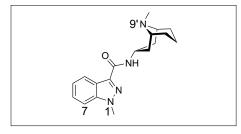
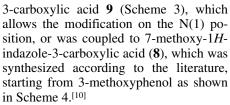


Fig. 2. Granisetron can be modified at the nitrogens N(1) and N(9') as well as C(7).

bled using a convergent synthetic strategy based on the bicyclic amine 5, an indazole core 8/9 and an aromatic diazirine acid 21. The bicyclic amine 5 was synthesized according to literature protocols (Scheme 2),^[8] starting from glutaraldehyde and aceton-1,3-dicarboxylic acid, which forms pseudopelletierine (3) in a Robinson tropinon-type synthesis.^[9] In this double-Mannich-decarboxylation cascade reaction bicyclic product 3 is produced in high yields (95%). The keto group was then converted to the hydroxylamine 4 which could be reduced with AlH₂ that was formed in situ from LiAlH₄ and H_2SO_4 , to yield *endo*amine 5 as a single isomer.

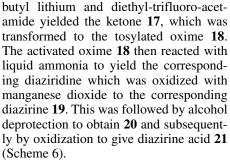
Primary amine **5** was coupled either to commercially available 1*H*-indazole-



After the introduction of a TMS group and conversion of the OTMS group to a better leaving group (OTf) the indazole core was formed by a 1,3-dipolar cycloaddition of *in situ* generated benzyne and ethyl diazoacetate. The subsequent saponification led to **8**, which could be coupled with bicyclic amine **5**, to yield **10**. The indazole nitrogen N(1) of compound **10** was methylated and the methoxygroup cleaved in the second step to allow further functionalization with linkers and the photoactivatable diazirine.

3-(*N*-Boc-amino)propyl bromide was used as a linker which was attached *via* the hydroxy group of **12**, or *via* the indazole nitrogen N(1) in the case of **11**.

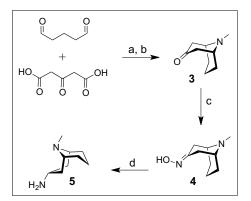
The subsequent deprotection gave the amines **13** and **14** (Scheme 5) which allowed the introduction of the photolabile diazirine group that was synthesized as shown in Scheme 6.^[11] Treating TBSprotected 4-bromo-phenylmethanol with



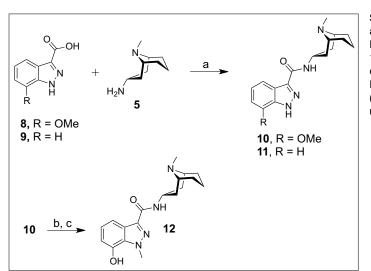
In the final reaction the granisetron derivatives 13 and 14 were coupled with the photolabile diazirine 21 to give the two photoaffinity probes 1 and 2 (Scheme 5).

Binding Affinity

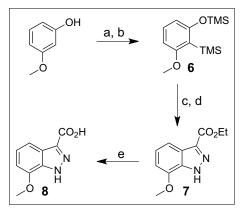
We explored the binding properties of the two compounds **1** and **2** at the 5-HT₃ receptor by competition with radiolabeled [³H]granisetron. Both compounds had binding affinities in the nanomolar range, with compound **2** showing the comparably higher affinity, with a K_i of 8 nM compared to **1** with a K_i of 123 nM.



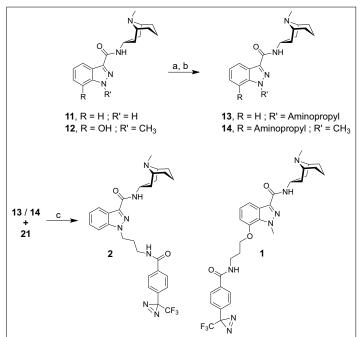
Scheme 2. Reagents and conditions: (a) MeNH₃Cl, Na₂HPO₄, NaOH, 20 h, r.t.; (b) HCl, 1 h, 100 °C, for both steps: 95%; (c) NaOAc, NH₃OHCl, 4 h, r.t. 92%; (d) H₂SO₄, LiAlH₄, THF, 8 h, r.t. to 45 °C, 30–92%.



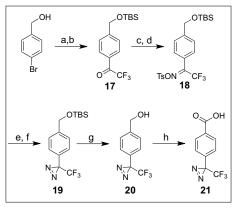
Scheme 3. Reagents and conditions: (a) HOBt, DCC, CH₂Cl₂, 14 h, r.t., 82% (**8**), quant. (**9**); (b) KOt-Bu, Mel, THF/DMF, 42%; (c) BBr₃, CH₂Cl₂, 24 h, r.t., crude.



Scheme 4. Reagents and conditions: (a) HMDS 1 h, 75 °C, crude; (b) *i*-Pr₂NH, *n*-BuLi, TMS-Cl, THF, 13 h, -78 °C to r.t., 87%; (c) *n*-BuLi, Tf₂O, Et₂O, 0 °C to r.t., 90%; (d) N₂CHCO₂Et, TBAF, THF, -78 °C to r.t., 24 h, 81%; (e) NaOH, MeOH, 14 h, 60 °C, quant.



Scheme 5. (a) KOt-Bu, 3-(N-boc-amino) propylbromide, THF/ DMF, 14 h, 0 °C to r.t.; (b) HCl, dioxane, 14 h, r.t., for both steps: 70% (**13**), 31% (**14**); (c) HATU, Et₃N, 24% (**2**), 20% (**1**).



Scheme 6. Reagents and conditions: (a) TBS-Cl, imidazole, DMF, 14 h, 40 °C, quant.; (b) *n*-BuLi, diethyl-trifluoro-acetamide, Et₂O, 4 h, -78 °C, 78%; (c) hydroxylamine HCl, Pyridine, EtOH, 4 h, 60 °C, crude; (d) TsCl, Pyridine, 3.5 h, both steps: 59%; (e) NH₃, Et₂O, 14 h, -78 °C to r.t., 90%; (f) MnO₂, CH₂Cl₂, 48 h, r.t. 70%; (g) HCl, MeOH 6 h, r.t., 53%; (h) KMnO₄, KOH, dioxane, 2 h, r.t., 70%.

In previous studies we showed that a bulky fluorophore attached to the indazole nitrogen of granisetron was tolerated without losing the binding affinity.^[7] As compounds **1** and **2** contained two identical linker/diazirine combinations at C(7) and N(1), their binding affinities can be directly compared. The data suggest that the granisetron binding pocket is less sterically confined around the N(1) nitrogen and thus this position is more likely to accommodate bulky modifications.^[12]

Irradiation of Diazirines

Diazirines are known to form a carbene by extruding N_2 when irradiated with lowenergy wavelength light (>300 nm).^[13] These long wavelengths and the ability of the newly formed highly reactive carbene to rapidly insert into C–H or heteroatom– H bonds, allows their use to covalently modify proteins.^[14]

First, we investigated the stability of our compounds in daylight conditions and their behavior under UV irradiation (365/302 nm) by dissolving them in deuterated methanol and measuring the ¹⁹F-NMR signal intensity at different irradiation times. We observed the formation of several products and identified two main compounds by MS, similar to those described by Hosoya and coworkers (Fig. 3).^[15] One was the insertion product **23** of the carbene into the D–O bond of d₄-MeOD, and the second was the isomerized diazo compound **22** shown in Scheme 7.

We further found that irradiating the sample with shorter wavelength light (302 nm) produced more insertion product **23** from **22**, presumably through the intermediacy of the carbene. This is in agreement with similar published studies.^[15]

To verify the stability of diazirines in lab light conditions we exposed 3-(4-(bromomethyl)phenyl)-3-(trifluoromethyl)-3*H*-diazirine to lab light and followed its decomposition with ¹⁹F-NMR. First photo-decomposition (16%) peaks similar to compounds **22** and **23** were detected after 6.5 h and full conversion of the diazirine was observed after 76 h. This low reactivity allows the handling of diazirine compounds without excessive light protection.

Conclusion

We have successfully synthesized two new photoaffinity probes based on the selective and high-affinity 5-HT₃ receptor antagonist granisetron. The probes were tested at the human 5-HT₃ receptor and revealed nanomolar binding. The probes were irradiated to investigate reaction times

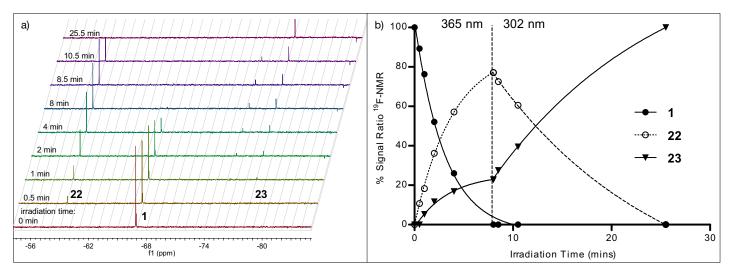
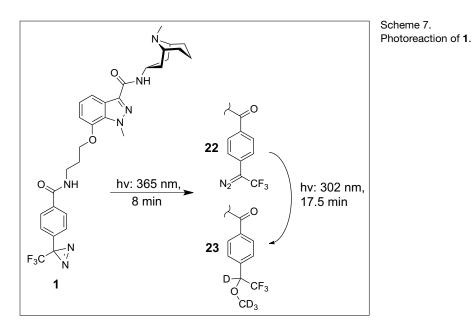


Fig. 3. a) ¹⁹F-NMR of irradiated diazirine **1** (–66.9 ppm) (irradiation: 8 min at 365 nm, then 17.5 min at 302 nm) b) ¹⁹F-NMR signal ratio of compound **1** to the two main decomposition products over irradiation time.



for further experiments in receptor labeling and showed a high reactivity at long wavelength irradiation. This makes them useful tools for investigating the 5-HT₂ receptor binding site, by photocrosslinking the probes with the receptor and subsequent digestion. The resulting fragments will be analyzed by mass spectrometry and, in conjunction with homology models and mutagenesis data, will help to identify the position and orientation of the probes in the receptor binding pocket. These experiments are part of our ongoing efforts to develop fluorescent and bioorthogonal molecular tools for the investigation of transmembrane proteins, including ligandgated ion channels (hERG), G-protein coupled receptors (adenosine receptors) and transporters (high-affinity glutamate transporters and TRPV channels).

Acknowledgements

The authors would like to thank Dr. Michele Leuenberger for support and many fruitful discussions. Metrohm AG and the Swiss Chemical Society are gratefully acknowledged for the award. This work was supported by the Swiss National Science Foundation (SNSFprofessorship PP00P2_123536 to M. L.). M.-D. R. was supported by a SNSF Sinergia grant (CRSII3-136222 to O.M.).

Received: January 16, 2014

[1] A. J. Thompson, S. C. Lummis, *Expert Opin. Ther. Targets* **2007**, *11*, 527.

- [2] J. Walstab, G. Rappold, B. Niesler, *Pharmacol. Ther.* 2010, 128, 146.
- [3] a) M. Pytliak, V. V. V. Mechírová, M. Felšöci, *Physiol. Res.* 2011, 60, 15; b) D. Hoyer, J. P. Hannon, G. R. Martin, *Pharmacol. Biochem. Behav.* 2002, 71, 533; c) M. Berger, J. A. Gray, B. L. Roth, *Annu. Rev. Med.* 2009, 60, 355.
- [4] N. M. Barnes, T. G Hales, S. C. R. Lummis, J. A. Peters, *Neuropharmacol.* **2009**, *56*, 273.
- [5] A. J. Thompson, H. A. Lester, S. C. R. Lummis, *Quart. Rev. Biophys.* 2010, 43, 449.
- [6] A. J. Thompson, S. C. R. Lummis, *Curr. Pharm. Des.* 2006, 12, 3615.
- [7] S. K. V.Vernekar, H. Y. Hallaq, G. Clarkson, A. J. Thompson, L. Silvestri, S. C. R. Lummis, M. Lochner, J. Med. Chem. 2010, 53, 2324.
- [8] O. Dold, K. Stach, W. Schaumann, US Patent Appl. No. 3.509.161, 1970.
- [9] R. Robinson, J. Chem. Soc., Trans. 1917, 111, 762.
- [10] Z. Liu, F. Shi, P. D. G. Martinez, C. Raminelli, R. C. Larock, J. Org. Chem. 2008, 73, 219.
- [11] a) L. B. Shih, H. Bayley, Anal. Biochem. 1985, 144, 132; b) J. H. Robson, J. Reinhart, J. Am. Chem. Soc. 1955, 77, 498; c) M. Nassal, Liebigs Ann. 1983, 9, 1510.
- [12] D. Kesters, A. J. Thompson, M. Brams, R. van Elk, R. Spurny, M. Geitmann, J. M. Villalgordo, A. Guskov, U. H. Danielson, S. C. R. Lummis, A. B. Smit, C. Ulens, *EMBO Rep* **2013**, *14*, 49.
- [13] L. Dubinsky, B. P. Krom, M. M. Meijler, *Bioorg. Med. Chem.* 2012, 20, 554.
- [14] A. Singh, E. R. Thornton, F. H. Westheimer J. Biol. Chem. **1962**, 237, 3006.
- [15] T. Hiramatsu, Y. Guo, T. Hosoya, Org. Biomol. Chem. **2007**, *5*, 2916.